This article was downloaded by:

On: 28 January 2011

Access details: Access Details: Free Access

Publisher Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713618290

N-Phospho- α -Amino Acids and Co-Evolution of Nucleic Acid and Protein Yu-Fen Zhao; Pei-Sheng Cao

To cite this Article Zhao, Yu-Fen and Cao, Pei-Sheng(1999) 'N-Phospho- α -Amino Acids and Co-Evolution of Nucleic Acid and Protein', Phosphorus, Sulfur, and Silicon and the Related Elements, 144: 1, 757 - 760

To link to this Article: DOI: 10.1080/10426509908546355 URL: http://dx.doi.org/10.1080/10426509908546355

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

N-Phospho-α-Amino Acids and Co-Evolution of Nucleic Acid and Protein

YU-FEN ZHAO and PEI-SHENG CAO

Department of Chemistry Bioorganic Phosphorus Chemistry Laboratory Tsinghua University Beijing, 100084, China

N-phospho- α -amino acids obtainable from aqueous condition were different from their corresponding β -analogues intrinsically. As the self-reproduction unites, N-phospho- α -amino acids were capable to produce non-random oligopeptides. As the phosphoryl donor, they were able to phosphorylate the nucleosides much faster than the deoxynucleosides.

Keywords: N-Phospho-α -amino acids; self-reproduction unites; non-random oligopeptides; peptide elongation; nucleotides elongation; molecular evolution

Non-Random Oligopeptides From N-phospho- a -amino acid

When the amino acids were produced through the prebiotic synthesis, they could be polymerized thermally, but the sequence of the oligopeptides were random. However, the N-phosphoamino acids¹ which were obtainable from the aqueous media, were able to produce a predictable peptides sequence. For example, when N-phospho-a-histidine incubated with serine at 40°C in aqueous media, the peptides His-ser, His-His-ser etc., were formed. While under the same condition, when N-phospho-a-serine incubated with histidine there were peptides Ser-His, Ser-Ser-His etc., produced. The N-phospho-a-amino acids were acted as the pentacoordinated "Charged Amino Acids" to

activate and initiate the polymerization reaction. In which the activated carboxylic center could be attacked by any nucleophile such as the amino group of an amino acid or peptide to produce the new peptides. It is indicated that the elongation of the peptide bonds was operated by repetition of this specific reaction for α -amino acids. Hence, the intrinsic chemical principle beheld by the N-phospho- α -amino acids is the natural endowment which N-Phospho β -amino acid did not have such properties.

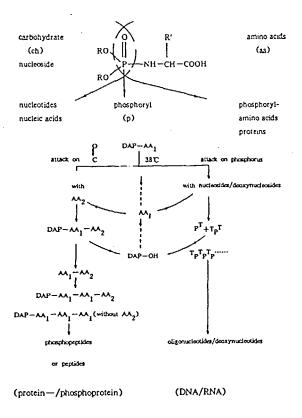
Nucleotides From N-phospho a -amino acids

The prebiotic synthesis of purines is robust; pyrimidine synthesis is satisfied, However, the nucleosides prebiotic synthesis is more difficult. For example, under the UV and the presence of phosphate or ethylmetaphosphate, the nucleosides could be produced from bases and ribose or deoxyribose².

Fox³ also showed that by treatment with polyphosphoric acid at 273K the nucleosides but not the deoxynucleosides could be converted into the nucleotides. Schwartz isolated the 2', 3' and 5'-AMP by mixing the adenosine with linear polyphosphoric acid⁴. These results demonstrated that the nucleosides were much easier to be phosphorylated than the deoxynucleosides.

In our laboratory, it was found that N-phospho- a -amino acids not only could self-assemble into oligopeptides, but also could phosphorylate the nucleosides or the deoxynucleosides to give mononucleotide, dinucleotides and their aminoacyl-conjugate³. The P-31-NMR HPLC, FAB-MS, CE(Capillary Electrophoresis). CE-MS were used for multiple checks on the products. These important basic properties were only for N-phospho- a -amino acids. The corresponding

N-phospho- β -amino acids did not possess the phosphorylation ability. In conclusion, the N-phospho- α -amino acids as the combination between



Scheme: N-phospho- a -amino acids and co-evolution of nucleic acid and protein

the limited nucleic acids and phosphorylated protein possess the autocatalysis ability to excute the self-reproduction and self-replication reactions which were paralleled to the contempory biosynthesis. And the oligopeptides and nucleotides were derived from N-phospho-aamino acids simultaneously, in a predictable non-random sequence⁶.

Acknowledgements

This work was supported by the National Natural Science Foundation of China, National Science and Technology Committee of China, Chinese National Education Ministry Committee and Tsinghua University.

References

- [1] Ji.G.J., Xue, C.B., Zeng, J.N., Li, L.P., Chai, W.G. and Zhao, Y.F., Synthesis, 6, 444(1988).
- [2] Ponnamperuma, C., Nature, 203, 400(1964).
- [3] Fox, S.W. and Harada, K., Arch. Biochem. Biophys. 86, 281(1960).
- [4] Schwartz, A., Nature, 218, 443 (1968).
- [5] Zhou, W.H., Ju, Y., Zhao, Y.F., Wang, Q.G., and Luo, G.A.:, Origins of Life and Evolution of the Biosphere, 279–285(1996).
- [6] Zhao, Y.F. and Cao, P.S., J. of Biological Physics, 20, 283(1994).
- [7] Zhao, Y.F. and Cao, P.S., J. Chela-Flores and F. Raulin (eds.), Chemical Evolution: Physics of the Origin and Evolution of Life, 279–285, 1996.